Structure Analysis Experiment Using Ultrasonic Waves

Introduction

This is the crystal structure of NaCl.

We usually measure the distance between Na and Cl(I call it *d*) by using X-rays.



But I was not able to use X-rays because it was too dangerous and expensive for me.

I thought I could measure *d* by using sound waves because X-rays and sound waves are both waves.



sonic waves

ultrasonic waves

ised <mark>ultrasonic waves</mark>

The wavelength of ultrasonic waves is short among sound waves. The experiment should not be obstructed by noise.

I was able to experiment something similar to it without using X-rays.

When Blag's equation is established, the two reflected waves are the same.

Principle of the structure analysis

Application of Blag's equation $2d \sin \theta = n \lambda (n = 1, 2, 3...)$

d:Distance *θ*:Angle of incidence

λ:Wavelength

Experiment

I made ultrasonic waves go into the model which is likened to the crystal of the NaCl(lined nails).

I changed the angle of incidence and measured the intensity of the reflected waves.



Instruments

An ultrasonic waving machine (*f*=40kHz)
Model of the crystal of the NaCl(*d*=1.0 × 10-3 m)
A protractor

$$v = 331.5 + 0.6t$$
 $t = 15^{\circ}C$ $v = 3.4 \times 10^{2} [m/s]$

v:Velocity *t*:Temperature *f*:Frequency

λ:Wavelength

$$\lambda = \frac{v}{f} = 8.5 \times 10^{-3} [m]$$

8.5×10 [m]

The model of NaCl

Conclusion

It is also possible to measure *d* in a model experiment using ultrasonic waves.



From Blag's equation,

• at θ =25° 52.5° are the strongest.

From this graph, the reflected waves

•From θ =25° d=1.0 × 10⁻³m •From θ =52.5° d=1.1 × 10⁻³m

•The real *d*=1.0 × 10⁻³m

sound waves

When this equation is established, the

same waves make each other stronger.

d of NaCl

and the *d* of NaCl are far shorter than those of sound waves.

I had a model experiment using sound

waves because the wavelength of X-rays

